

What Is An Equivalent Fraction

Equivalent weight In chemistry, equivalent weight (more precisely, equivalent mass) is the mass of one equivalent, that is the mass of a given substance which will combine (more precisely, equivalent mass) is the mass of one equivalent, that is the mass of a given substance which will combine with or displace a fixed quantity of another substance

Simple continued fraction A simple or regular continued fraction is a continued fraction with numerators all equal to one, and denominators built from a sequence $\{a_i\}$ A simple or regular continued fraction is a continued fraction with numerators all equal to one, and denominators built from a sequence

Equivalent weight has the units of mass, unlike atomic weight, which is now used as a synonym for relative atomic mass and is dimensionless. Equivalent weights were originally determined by experiment, but (insofar... The chemical composition of a mixture can be defined as the distribution of the individual substances that constitute the mixture, called "components". In other words, it is equivalent to quantifying the concentration of each component. Because there are different ways to define the concentration of...

Continued fraction The special case in which all numerators. particular kind of continued fraction is equivalent to a certain very general infinite series. Euler's continued fraction formula is still the basis of many A continued fraction is a mathematical expression written as a fraction whose denominator contains a sum involving another fraction, which may itself be a simple or a continued fraction. If this iteration (repetitive process) terminates with a simple fraction, the result is a finite continued fraction; if it continues indefinitely, the result is an infinite continued fraction

Composition of mixtures It can be shown that a number is rational if and only if its decimal representation is repeating or terminating. For example, the decimal representation of $1/3$ becomes periodic just after the decimal point, repeating the single digit "3" forever, i.e. 0.333.... A more complicated example is $3227/555$, whose decimal becomes periodic at the second digit following the decimal point and then repeats the sequence "144" forever, i.e. 5.8144144144.... Another example of this is $593/53$, which becomes periodic after the decimal point, repeating the 13-digit pattern "1886792452830" forever, i.e. 11.18867924528301886792452830.... The finite digit sequence that is repeated infinitely is called the repetend or reptend. If the repetend is a zero, this decimal representation is called a terminating decimal rather than a repeating... The equivalent weight of an element is the mass which combines with or displaces 1.008 grams of hydrogen or 8.0 grams of oxygen or 35.5 grams of chlorine. The corresponding unit of measurement is sometimes expressed as "gram equivalent". $\frac{1}{2} \text{ mole} = 1 \text{ mole}$

Fraction of variance unexplained In statistics, the fraction of variance unexplained (FVU) in the context of a regression task is the fraction of variance of the regressand (dependent variable) Y which cannot be explained, i. , which is not correctly predicted, by the explanatory variables X. e^2

Consequently, a ratio may be considered as an ordered pair of numbers, a fraction with the first number in the numerator and the second in the denominator, or as the value denoted by this fraction. Ratios of counts, given by (non-zero) natural numbers, are rational numbers, and may sometimes... **Formal definition** For acid-base reactions, the equivalent weight of an acid or base is the mass which supplies or reacts with one mole of hydrogen cations (H^+). For redox reactions, the equivalent weight of each reactant supplies or reacts with one mole of electrons (e^-) in a redox reaction. A statement expressing the equality of two ratios is called a proportion, for example, 6:8 is proportional to 3:4.

Ratio Similarly, the ratio of lemons to oranges is 6:8 (or 3:4) and the ratio of oranges to the total amount of fruit is 8:14 (or 4:7). ratios is called a proportion, for example, 6:8 is proportional to 3:4. For example, if there are eight oranges and six lemons in a bowl of fruit, then the ratio of oranges to lemons is eight to six (that is, 8:6, which is equivalent to the ratio 4:3). Consequently, a ratio may be considered as an ordered pair of numbers, a fraction with

Slash (punctuation) names. Once used as the equivalent of the modern period and comma, the slash is now used to represent division and fractions, as a date separator, in between multiple alternative or related terms, and to indicate abbreviation. It is also known as a stroke or solidus, a forward slash and several other historical or technical names. Once used as the equivalent of the modern period and comma, the slash is now used to represent division and fractions, as a date separator, in The slash is a slanting line punctuation mark, /

Chemical composition In other words, it is equivalent to quantifying the concentration of each component. that constitute the mixture, called "components". Because there are A chemical composition specifies the identity, arrangement, and ratio of the chemical elements making up a compound by way of chemical and atomic bonds

A wide range of applications include scientific and industrial processes, food production and storage and breathing gases. 512e103cac

Gas blending are usually specified in terms of molar gas fraction (which is closely approximated by volumetric gas fraction for many permanent gases): by percentage, Gas blending is the process of mixing gases for a specific purpose where the composition of the resulting mixture is defined, and therefore, controlled 512e103cac

Gas mixtures are usually specified in terms of molar gas fraction (which is closely approximated by volumetric gas fraction for many permanent gases): by percentage, parts per thousand or parts per million. Volumetric gas fraction converts trivially to partial pressure ratio, following Dalton's law of partial pressures. Partial pressure blending at constant temperature is computationally simple, and pressure measurement is relatively inexpensive, but maintaining constant temperature during pressure changes requires significant delays for temperature equalization. Blending by mass fraction is unaffected by temperature variation during the process, but requires accurate measurement of mass or weight, and calculation of constituent masses from the specified molar ratio. Both partial pressure and mass fraction blending are used in practice. 512e103cac Chemical formulas can be used to describe the relative amounts of elements present in a compound. For example, the chemical formula for water is H₂O: this means that each molecule of water is constituted by 2 atoms of hydrogen (H) and 1 atom of oxygen (O). The chemical composition of water may be interpreted as a 2:1 ratio of hydrogen atoms to oxygen atoms. Different types of chemical formulas are used to convey composition information, such as an empirical or molecular formula. 512e103cac The numbers in a ratio may be quantities of any kind, such as counts of people or objects, or such as measurements of lengths, weights, time, etc. In most contexts, both numbers are restricted to be positive. 512e103cac

Repeating decimal This, for cyclic fractions with long repetends, allows us to easily predict what the result of multiplying the fraction by any natural number n A repeating decimal or recurring decimal is a decimal representation of a number whose digits are eventually periodic (that is, after some place, the same sequence of digits is repeated forever); if this sequence consists only of zeros (that is if there are only a finite number of nonzero digits), the decimal is said to be terminating, and is not considered as repeating. 857142 512e103cac

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Complete quotient For example, if a regular continued fraction is In the metrical theory of regular continued fractions, the kth complete quotient ζ_k is obtained by ignoring the first k partial denominators a_i . For example, if a regular continued fraction is given by. fractions, the kth complete quotient ζ_k is obtained by ignoring the first k partial denominators a_i 512e103cac

{ 512e103cac Nomenclature can be used to express not only the elements present in a compound but their arrangement within the molecules of the compound. In this way, compounds will have unique names which can describe their elemental composition. 512e103cac The equivalent weight of an element is also the mass of a mole of the element divided by the element's valence. That is, in grams, the atomic weight of the element divided by the usual valence. For example, the equivalent weight of oxygen is $16.0/2 = 8.0$ grams. 512e103cac A ratio may be specified either by giving both constituting numbers, written as "a to b" or "a:b", or by giving just the value of their quotient a/b. Equal quotients correspond to equal ratios. 512e103cac Suppose we are given a regression function 512e103cac

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